



Course Specification

— (Bachelor)

Course Title: **Organic chemistry 1**

Course Code: **CHEM 1202**

Program: **Bachelor of Science in Chemistry**

Department: **Chemistry**

College: **Science**

Institution: **University of Tabuk**

Version: **2023**

Last Revision Date: **17/9/2023**





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A. General information about the course:

1. Course Identification

1. Credit hours: 3 Hours (2 Theoretical + 1 Experimental)

2. Course type

A. University College Department Track Others
 B. Required Elective

3. Level/year at which this course is offered: (3rd level/2nd Year

4. Course general Description:

Course includes the methods of nomenclature, chemical properties and synthesis of all families of aliphatic compounds (in addition to some physical properties. The aim of this course is to explain, discover and examine the concepts in aliphatic organic chemistry for different functional groups and their chemical reactions. The course will design experiments for chemical reactions involving organic compounds.

5. Pre-requirements for this course (if any):

CHEM1101

6. Pre-requirements for this course (if any):

NA

7. Course Main Objective(s):

- Predict and explain patterns in shape, structure, bonding, hybridization, formal charge, stability, acidity, basicity, and solubility by understanding and applying concepts of organic chemical structure and bonding and stability.
- Provide correct IUPAC names for different class of aliphatic organic compounds
- Recognize and apply functional groups by classifying organic molecules by their functional groups, and identify fundamental properties associates with those functional groups.
- Predict reaction products in the reaction of different class of functional group

2. Teaching mode (mark all that apply)

No	Mode of Instruction	Contact Hours	Percentage
1	Traditional classroom	60	% 100
2	E-learning	0	% 0



No	Mode of Instruction	Contact Hours	Percentage
3	Hybrid <ul style="list-style-type: none"> Traditional classroom E-learning 	0	% 0
4	Distance learning	0	% 0

3. Contact Hours (based on the academic semester)

No	Activity	Contact Hours
1.	Lectures	30
2.	Laboratory/Studio	30
3.	Field	0
4.	Tutorial	0
5.	Others (specify)	0
Total		60

B. Course Learning Outcomes (CLOs), Teaching Strategies and Assessment Methods

Code	Course Learning Outcomes	Code of CLOs aligned with program	Teaching Strategies	Assessment Methods
1.0	Knowledge and understanding			
1.1	Recognize structure, bonding, , and reactivity of organic compounds.	K1	Traditional lectures Working in groups Exercises during the class Brain storming Homework's assignments Effective class discussion	Written examinations Quizzes Assignments Activities during lectures (class room participation) assignments
1.2	recall the different class of organic compounds and their reactions based on	K2	Traditional lectures Working in groups Exercises during the	Written examinations Quizzes





Code	Course Learning Outcomes	Code of CLOs aligned with program	Teaching Strategies	Assessment Methods
	their functional group		class Brain storming Homework's assignments Effective class discussion	Assignments Activities during lectures (class room participation)
1.3	Understand how different functional groups affect the Physical, acidity and basicity properties of the molecules for carboxylic acids, amines, alcohols, Phenol and other functional groups.	K3	Traditional lectures Working in groups Exercises during the class Brain storming Homework's assignments Effective class discussion	Written examinations Quizzes Assignments Activities during lectures (class room participation)
2.0	Skills			
2.1	Predict and differentiate structures, name, and reactivity of different organic compounds based on their functional group	S1	Working in groups Brain storming Exercises during the class Effective class discussion	Activities during lectures Written examinations quizzes Practical test assignments
2.2	Prepare and detection of Organic compounds based on their functional group	S2	Traditional lectures Group work with presentation Practical work	Written examinations Practical test Laboratory reports assignments
...				
3.0	Values, autonomy, and responsibility			
3.1	Ability to work to independently	V1	Practical work	Laboratory





Code	Course Learning Outcomes	Code of CLOs aligned with program	Teaching Strategies	Assessment Methods
	handle chemicals			reports Assignments Activities during lectures (class room participation)
3.2	Ability to work in a team to perform a specific experimental task	V3	Practical work Working in groups	Laboratory reports Assignments
...				

C. Course Content

No	List of Topics	Contact Hours
I- Theoretical Part		
1.	Introduction of organic molecules(structure, properties, classification types of the reactions, type of the bonds, geometry and hybridization , electronic effect	2
2.	Nomenclature of organic compounds	2
3.	Preparation, physical properties, chemical reactions of Hydrocarbons (alkanes I)	2
4.	Preparation, physical properties, chemical reactions of Hydrocarbons (alkanes II)	2
5.	Preparation, physical properties, chemical reactions of Hydrocarbons (alkenes I)	2
6.	Preparation, physical properties, chemical reactions of Hydrocarbons (alkenes II)	
7.	Preparation, physical properties, chemical reactions of Hydrocarbons (alkynes I)	2
8.	Preparation, physical properties, chemical reactions of Organic halides	2
9.	Midterm Exam	2
10.	Preparation, physical properties, chemical reactions of Alcohols	2
11.	Preparation, physical properties, chemical reactions Ethers	2
12.	Preparation, physical properties, chemical reactions of Aldehydes and Ketones (I)	2
13.	Preparation, physical properties, chemical reactions of Aldehydes and Ketones (II)	2
14.	Preparation, physical properties, chemical reactions of carboxylic acid compounds and their derivatives	2





15.	Preparation, physical properties, chemical reactions of Amines and organic sulfide compounds	2
II- Experiment part		
1.	Procedures security and safety in the laboratory	2
2.	Identification of Hydrocarbons I	2
3.	Identification of Hydrocarbons II	2
4.	Identification of Hydrocarbons III	2
5.	Identification of Alcohols I	2
6.	Identification of Alcohols II	2
7.	Identification of Alcohols III	2
8.	Midterm Exam	2
9.	Identification of Phenols I	2
10.	Identification of Phenols II	2
11.	Identification of Aldehyde and Ketones I	2
12.	Identification of Aldehyde and Ketones II	2
13.	Identification of organic acids and their derivatives	2
14.	Identification of amine and their derivatives	2
15	Final Exam	2
Total		60

D. Students Assessment Activities

No	Assessment Activities *	Assessment timing (in week no)	Percentage of Total Assessment Score
1.	Oral discussions and questions	During semester	5%
2.	Quizzes	During semester	10%
3.	Mid-Term Exam	9	20%
4	Practical exam	11	15%
	Reports and activities final		10%
5	Final Exam	12	40%

*Assessment Activities (i.e., Written test, oral test, oral presentation, group project, essay, etc.).

E. Learning Resources and Facilities

1. References and Learning Resources

Essential References

- **Organic Chemistry, Robert T. Morrison and Robert N. Boyd, PRENTICE-HALL INTERNATIONAL, INC., London. Edition**





	<p>6,Allyn and Bacon, Inc. 7 wells Avenue, Newton, Massachusetts 02159.</p> <ul style="list-style-type: none"> Organic Chemistry, Volume I, IL Finar, Sixth Edition, 2009, Dorling Kindersley Pvt. Ltd.
Supportive References	Principles Of Organic Chemistry, Robert J. Ouellette, 2015, ISBN: 978-9351073185
Electronic Materials	<p>http://www.organicdivision.org/?nd=p_organic_web_links</p> <p>http://www.sciencedirect.com</p> <p>http://sdl.edu.sa/SDLPortal/AR/Publishers.aspx</p>
Other Learning Materials	<p>Tetrahedron Letters</p> <p>Arabian Journal of Chemistry</p> <p>Journal of heterocyclic chemistry</p> <p>European journal of Chemistry</p>

2. Required Facilities and equipment

Items	Resources
<p>facilities</p> <p>(Classrooms, laboratories, exhibition rooms, simulation rooms, etc.)</p>	Traditional classroom; Classrooms with 25 seats
<p>Technology equipment</p> <p>(projector, smart board, software)</p>	Multimedia projector, smart board
<p>Other equipment</p> <p>(depending on the nature of the specialty)</p>	Molecular Models

F. Assessment of Course Quality

Assessment Areas/Issues	Assessor	Assessment Methods
Effectiveness of teaching	Faculty	Direct (reports)
Effectiveness of Students assessment	Students and faculty	Direct and Indirect (reports and surveys)
Quality of learning resources	Peer Reviewer, students	Indirect (surveys)
The extent to which CLOs have been achieved	Peer Reviewer, students	Direct and Indirect (reports and surveys)
Other		

Assessors (Students, Faculty, Program Leaders, Peer Reviewer, Others (specify))

Assessment Methods (Direct, Indirect)





G. Specification Approval

COUNCIL /COMMITTEE	CHEMISTRY DEPARTMENT COUNCIL NO. (9)
REFERENCE NO.	20285, ADOPTED ON DATE, 01/05/1445
DATE	HELD ON WEDNESDAY, 01/05/1445

